



## Making Light Work

Dynamic Light Scattering  
Electrophoretic Light Scattering  
Static Light Scattering

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## Contents



- Instruments using light
- Light scattering
- Nanoparticle Tracking Analysis (NTA)
- Dynamic light scattering (DLS)
- Electrophoretic light scattering (ELS)
- Static light scattering (SLS)

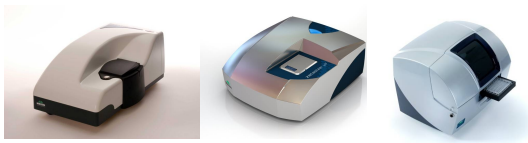
# Contents



- **Instruments using light**
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- Electrophoretic light scattering (ELS)
- Static light scattering (SLS)

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# Instruments Using Light



**Zetasizer Range**  
**(DLS, ELS, SLS)**



**OMNISEC**  
**SLS (LALS, RALS, MALS)**



**Nanosight**  
**(NTA)**



**Mastersizer**  
**SLS**  
**(LALLS)**

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## Light Scattering



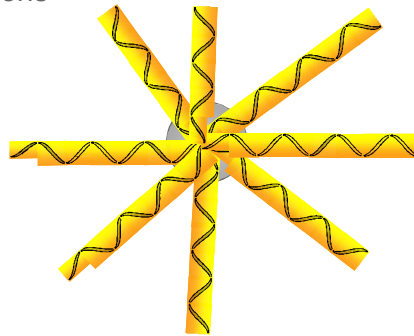
- The phenomenon of light scattering can be used in multiple ways to characterise a macromolecule/particle in solution/suspension
- A photon from an incident beam is absorbed by a macromolecule/particle and re-emitted in all directions

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## Light Scattering



- The phenomenon of light scattering can be used in multiple ways to characterise a macromolecule/particle in solution/suspension
- A photon from an incident beam is absorbed by a macromolecule/particle and re-emitted in all directions



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## Rayleigh Theory



- Applicable for small particles and molecules whose diameters are less than  $1/10^{\text{th}}$  of the laser wavelength ( $\lambda$ )
- He-Ne laser ( $\lambda = 633\text{nm}$ )  $\approx < 60\text{nm}$
- Isotropic scattering i.e. equal in all directions
- **Intensity =  $d^6$**

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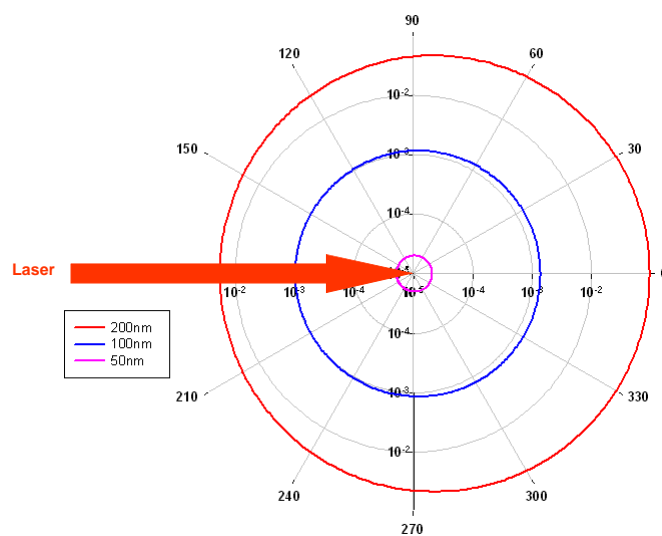
## Mie Theory



- Exact description of how spherical particles of all sizes and optical properties scatter light
- Particles  $> \lambda/10$ , scattering distorts in the forward scattering direction
- Particle size  $\Rightarrow \lambda$ , scattering is a complex function of maxima and minima with respect to angle which is correctly explained by Mie theory

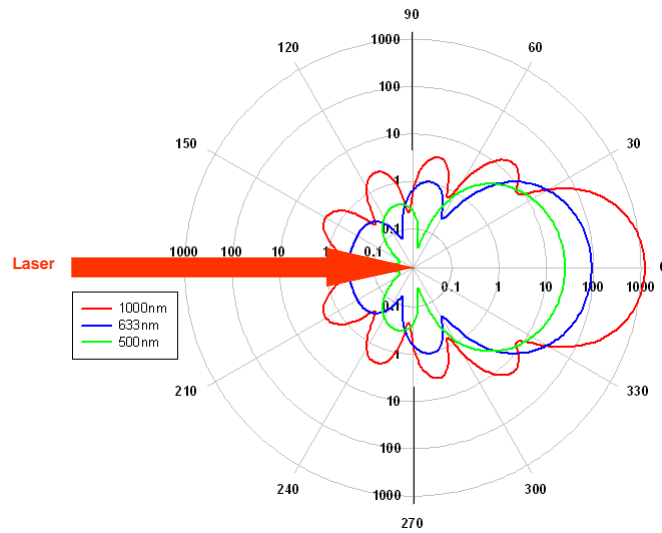
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## Mie Theory: Polar Plots



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## Mie Theory: Polar Plots



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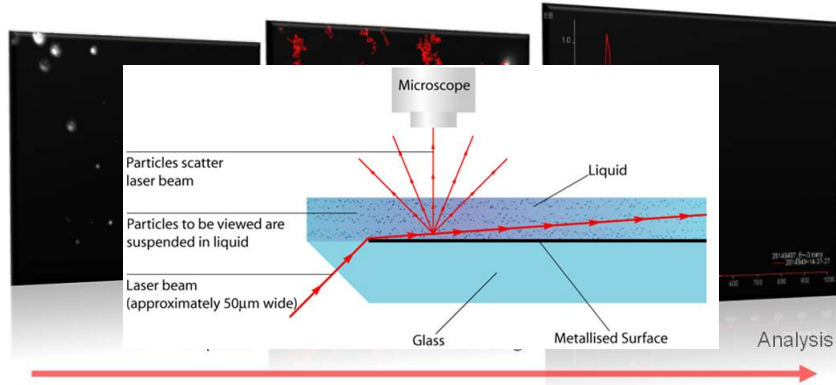
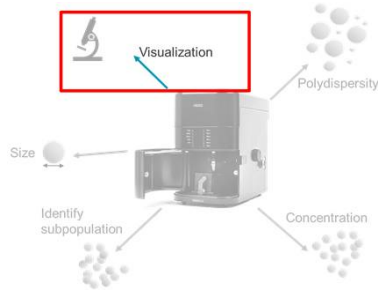


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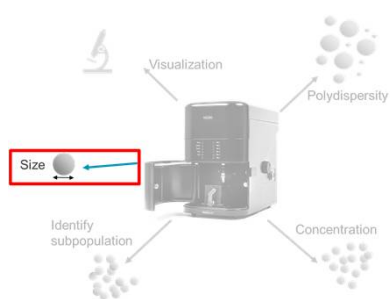
## Visualization

### Nanoparticle Tracking Analysis (NTA)



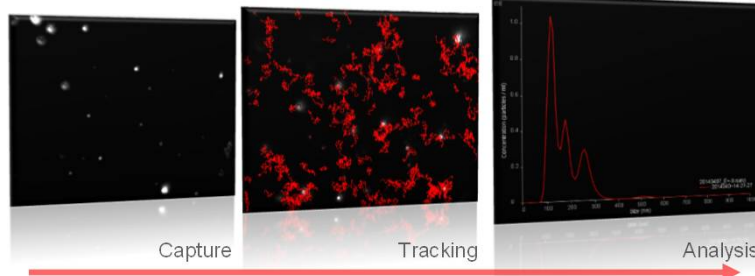
No need to know sample refractive index or density

## Size Measurement

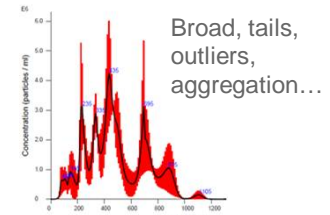
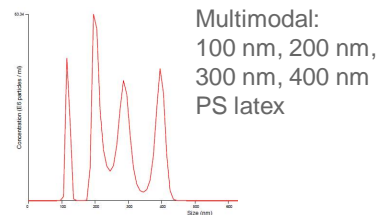
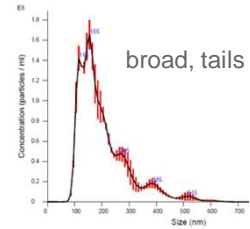
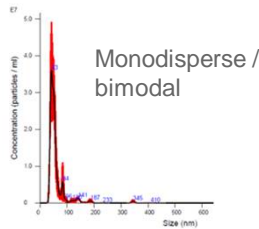
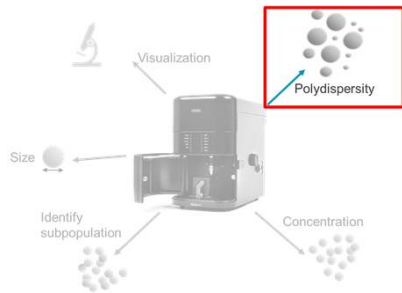


- NanoSight tracks nanoparticles moving under Brownian Motion
- Diffusion coefficients determined by tracking the particles and used to calculate size ([Stokes-Einstein equation](#))
- Smaller particles move faster than larger particles

ISO19430: Particle Tracking Analysis (PTA) method describes limitations, quantification parameters as well as instructions on how to operate the equipment in a certified manner.



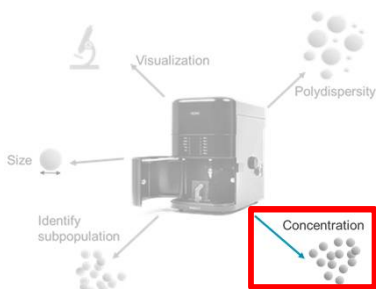
## Polydisperse Samples Sample distribution examples



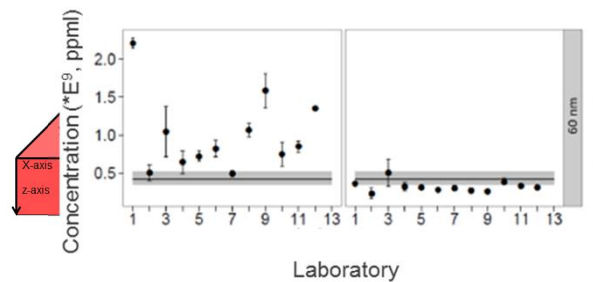
15 Nanoparticle Tracking Analysis (NTA)

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## Concentration



- NanoSight tracks individual particles, allowing their exact number to be determined
- The volume is determined by the field of view (X and Y) and from the laser beam profile (Z)



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16 Nanoparticle Tracking Analysis (NTA)

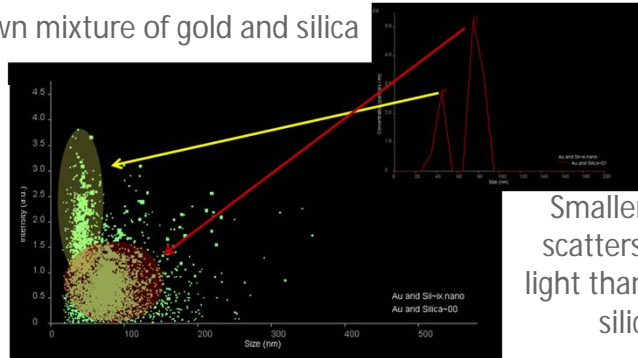
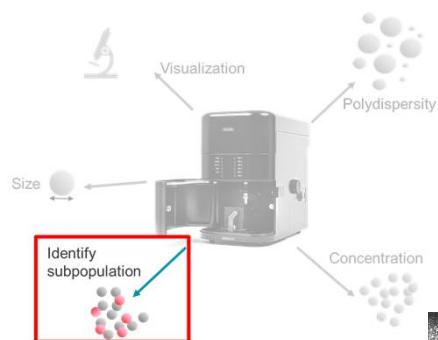


## Identify Subpopulation

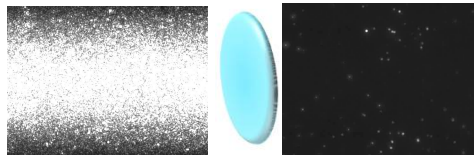
Relative Refractive Index or Fluorescence



Unknown mixture of gold and silica



Smaller gold scatters more light than bigger silica



Different laser wavelengths for different fluorophores

- 1.405 (430) nm
- 2.488 (500) nm
- 3.532 (565) nm
- 4.642 (650) nm

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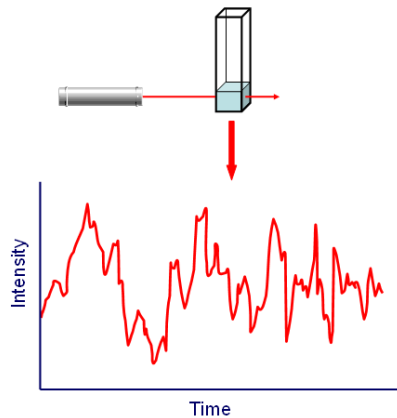


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## Dynamic Light Scattering

### Particle and Molecular Size Measurements



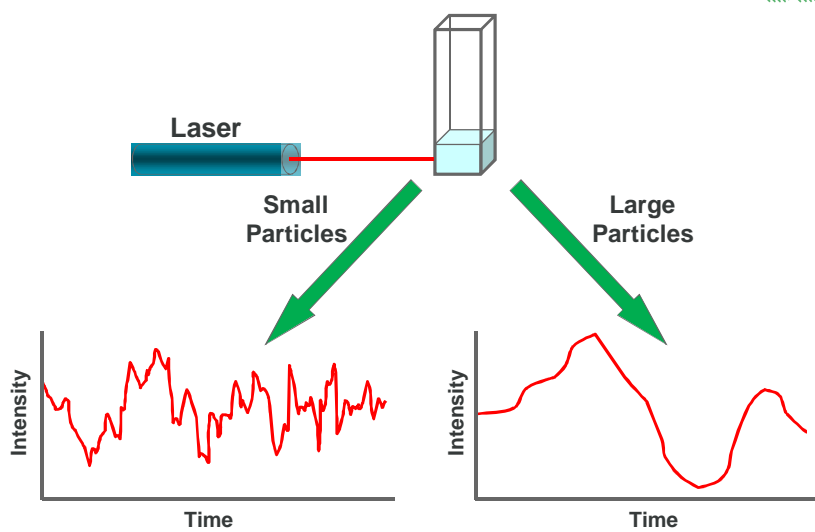
**Time-dependent fluctuations** in the scattering intensity to determine the **translational diffusion coefficient (D)**, and subsequently the **hydrodynamic diameter (D<sub>H</sub>)** (from the Stokes-Einstein equation)

$$D = \frac{kT}{3\pi\eta D_H}$$

Where k = Boltzmann's constant,  
T = absolute temperature  
η = viscosity

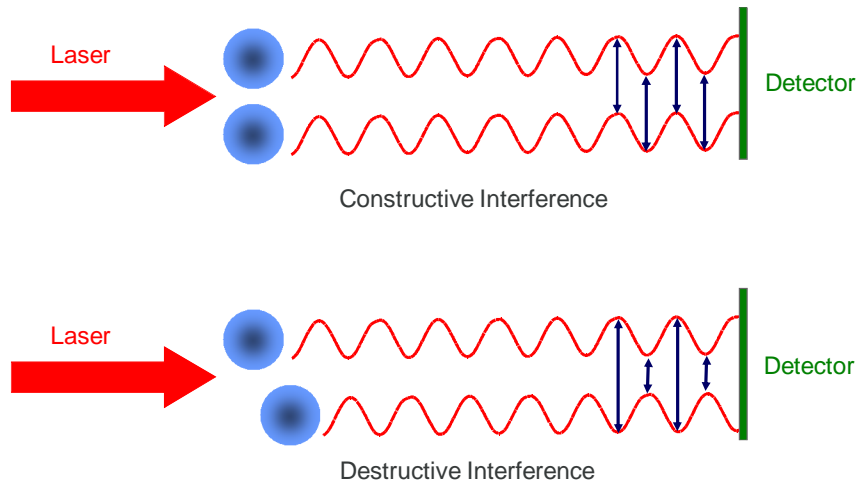
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## Brownian Motion and Intensity Fluctuations



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## Constructive and Destructive Interference



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## Correlation in Dynamic Light Scattering



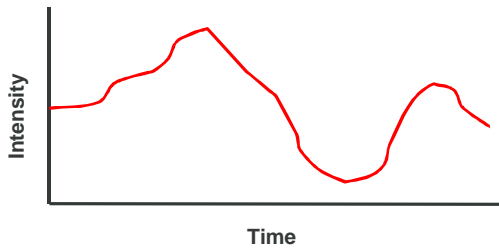
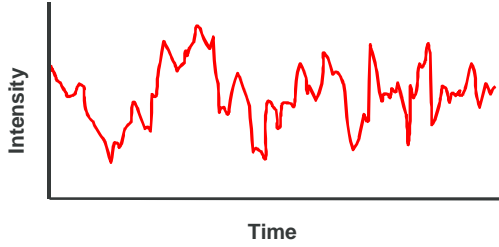
- Technique for extracting the time dependence of a signal in the presence of “noise”
- Time analysis carried out with a correlator
- Constructs the time autocorrelation function  $G(\tau)$  of the scattered intensity according to

$$G(\tau) = \left\langle \frac{I(t_0) * I(t_0 + \tau)}{I(t_\infty)^2} \right\rangle$$

where  $I$  = intensity,  $t$  is the time and  $\tau$  = the delay time

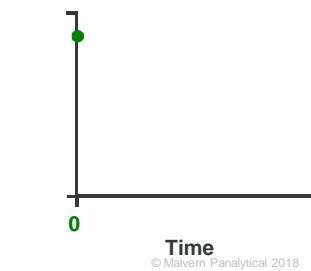
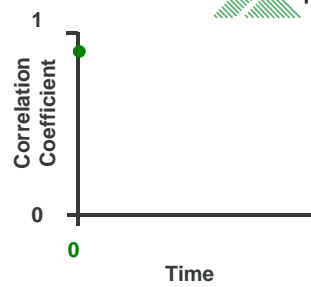
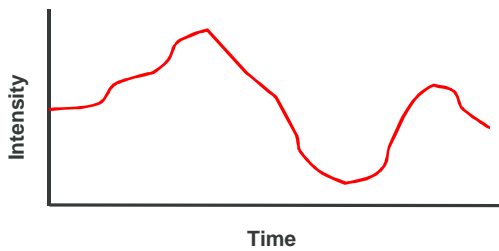
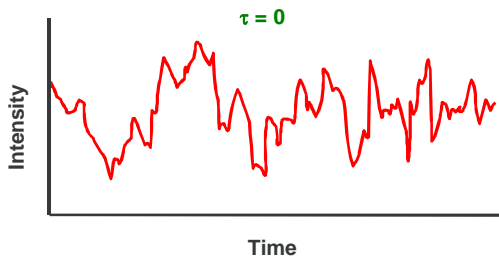
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### Correlation



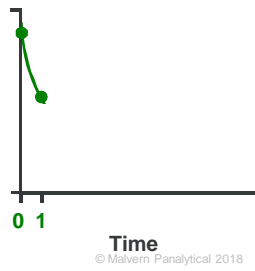
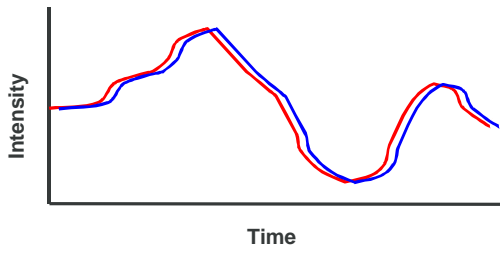
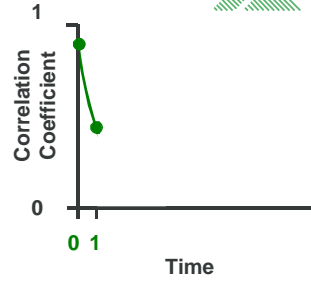
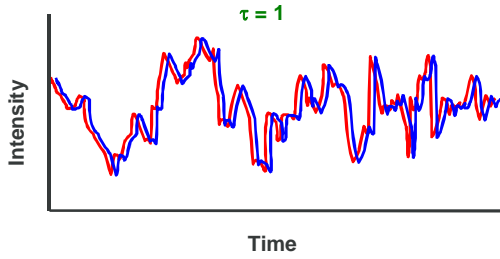
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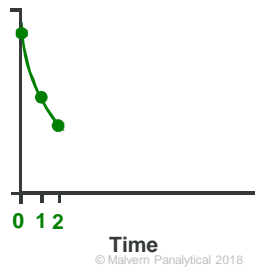
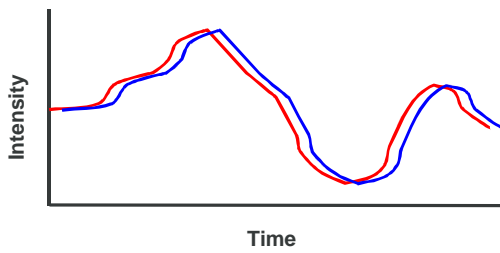
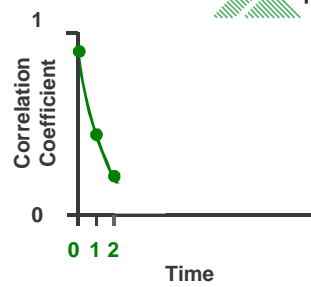
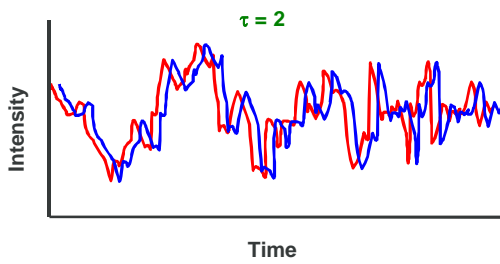


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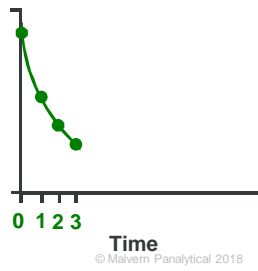
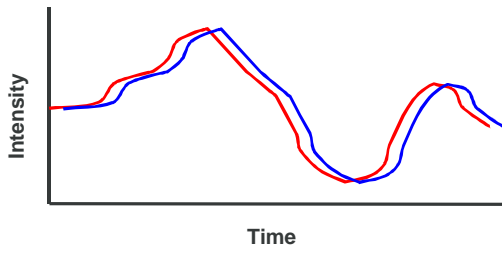
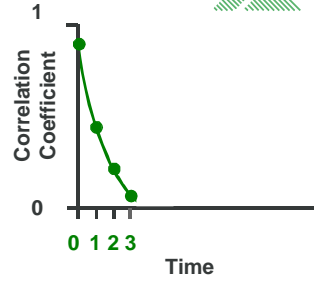
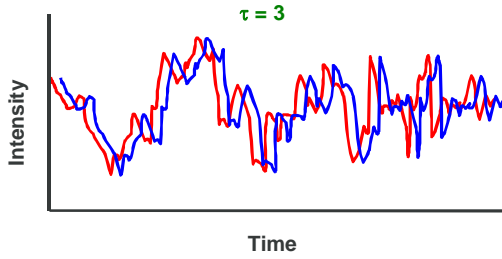
### Correlation



### Correlation

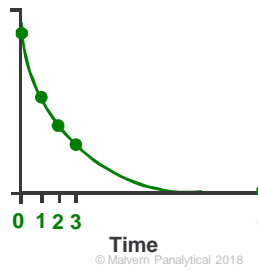
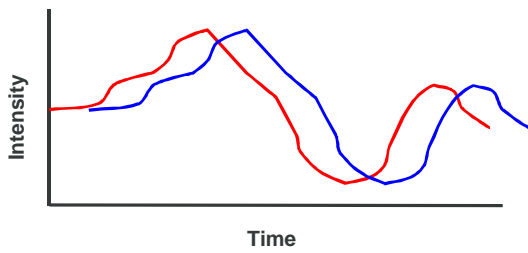
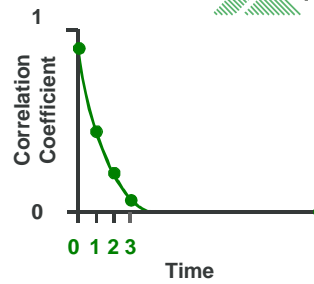
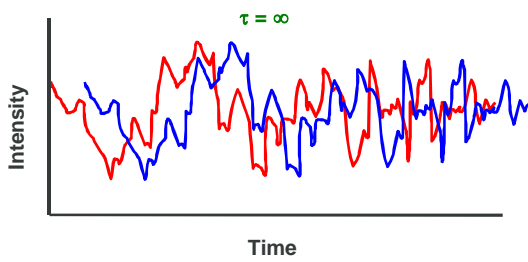


### Correlation



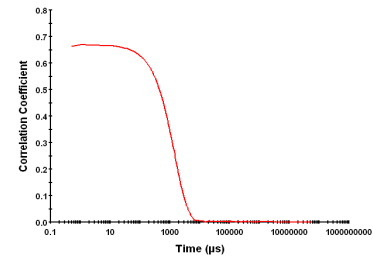
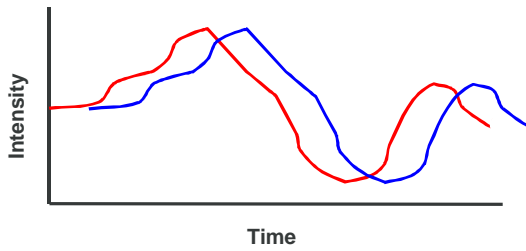
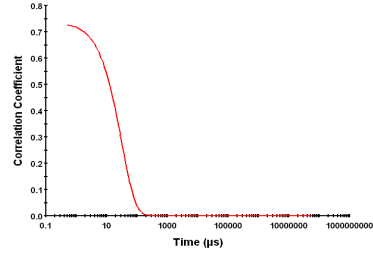
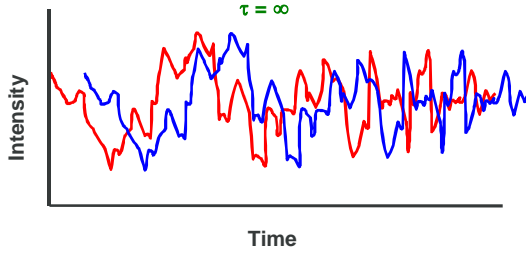
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### Correlation



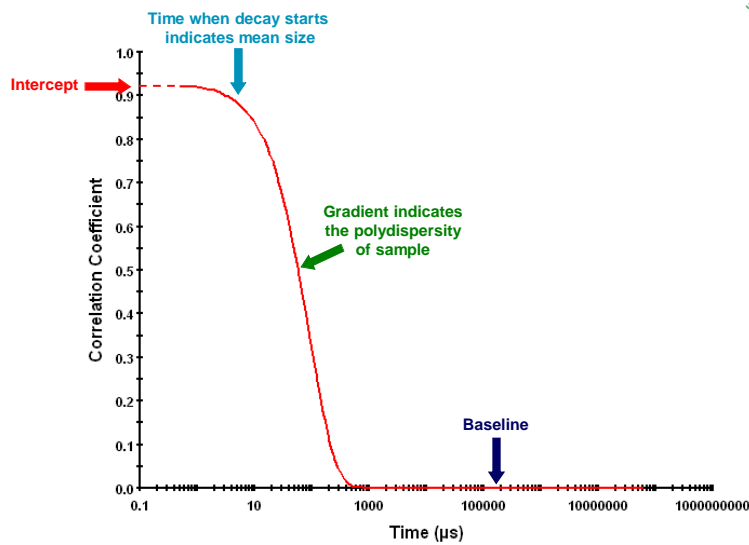
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### Correlation



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### Correlation Functions



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## Analysing The Correlation Function



- Correlation function contains the diffusion coefficient information required to be entered into the Stokes-Einstein equation
- The diffusion coefficients are obtained by fitting the correlation function with a suitable algorithm

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## Analysing The Correlation Function



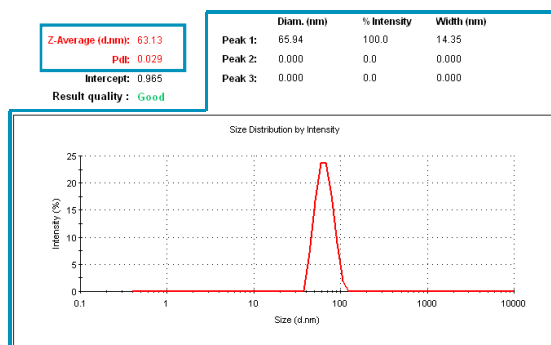
Two different analyses are performed:

### Cumulants analysis

- Mean size (z-average)
- Polydispersity index

### Distribution analysis

- Distribution of sizes



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## Cumulants Analysis



- Defined in the International Standards [ISO22412 \(2017\)](#)
- Only gives a [mean particle size \(z-average\)](#) and an [estimate of the width of the distribution \(polydispersity index\)](#)
- Only the dispersant refractive index and viscosity are required for this analysis

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## z-Average Diameter



- Definition of the z-Average Diameter ( $Z_D$ ):

**The intensity-weighted mean diameter derived from the cumulants analysis**

- Specific to light scattering
- Very sensitive to the presence of aggregates or large contaminants due to the inherent intensity weighting

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## Polydispersity Index



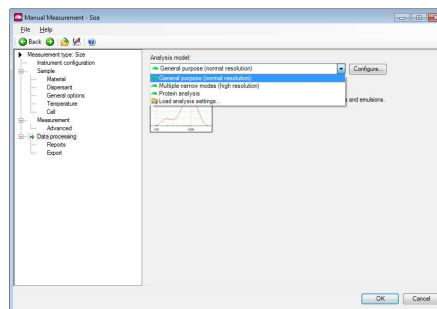
- Definition of the Polydispersity Index (Pdl):

**A dimensionless measure of the broadness of the size distribution calculated from the cumulants analysis**

- Ranges from 0 to 1 in the Zetasizer software
- Values > 1 indicate that the distribution is so polydisperse, the sample may not be suitable for measurement by DLS

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## Distribution Algorithms



- **General Purpose** (non-negative least squares (NNLS))
- **Multiple Narrow Modes** (non-negative least squares (NNLS))
- **Protein Analysis** (L-curve)
- The difference between these algorithms is their resolution capability

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## Zetasizer Distribution Algorithms



- **General Purpose**

- Suitable for the majority of samples where no knowledge of the distribution is available
- Will give broad, smooth distributions

- **Multiple Narrow Modes**

- Suitable for samples suspected to contain discrete populations
- Will give narrow peaks

- **Protein Analysis**

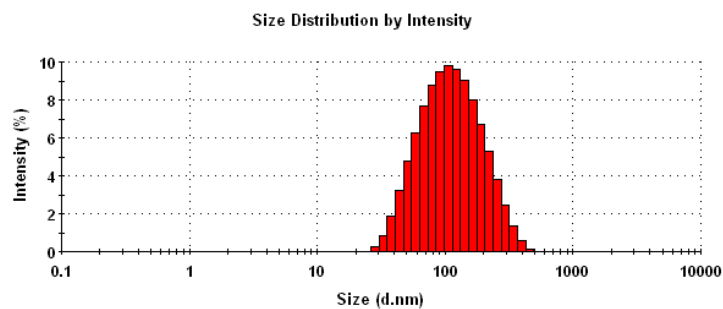
- Best suited for protein samples – will give narrow peaks
- Automatically picks the optimal distribution

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## Size Distributions in the Zetasizer Software



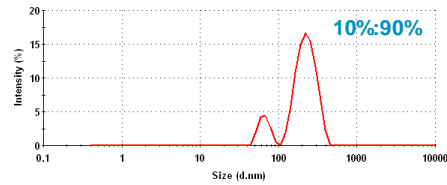
- Primary size distribution is **intensity**-weighted
- A plot of the **relative intensity of light scattered** by particles (on the Y axis) versus various **size classes** (on the X axis) which are logarithmically spaced



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## Intensity Size Distributions

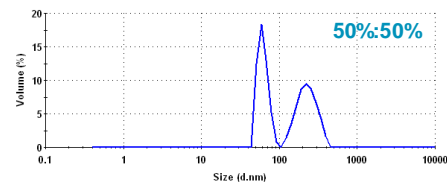
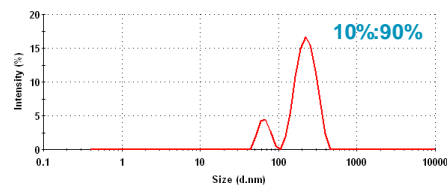
- Primary result
- Based upon the intensity of light scattered by particles
- Sensitive to the presence of large particles/aggregates /dust
- Only the dispersant viscosity and refractive index values are required



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## Volume Size Distributions

- Derived from the intensity distribution using Mie theory
- Equivalent to the mass or weight distribution
- Particle optical properties required to make this transformation
  - Particle refractive index
  - Particle absorption

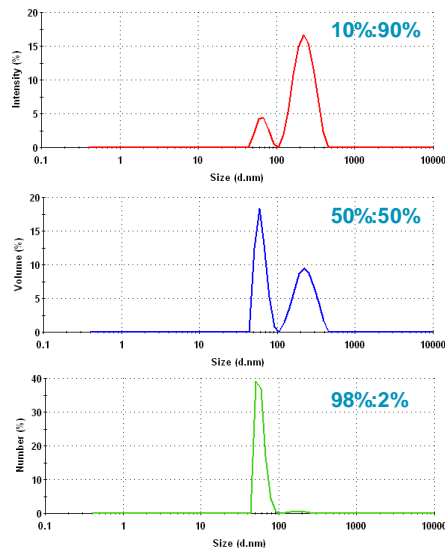


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## Number Size Distributions



- Derived from the intensity distribution using Mie theory
- Particle optical properties required to make this transformation
  - Particle refractive index
  - Particle absorption



## Size Distributions From DLS



- Transformation from intensity to volume or number makes the following assumptions:
  - All particles are spherical
  - All particles have an homogenous and equivalent density
  - The optical properties are known (refractive index and absorbance)
  - There is no error in the intensity distribution

## Size Distributions From DLS



- DLS technique tends to overestimate the width of the peaks in the distribution
- This effect is magnified in the transformations to volume and number
- The volume and number size distributions should only be used for estimating the **relative amounts** of material in separate peaks as the means and particularly the widths are less reliable

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## Volume/Number Distributions

### Recommended Use

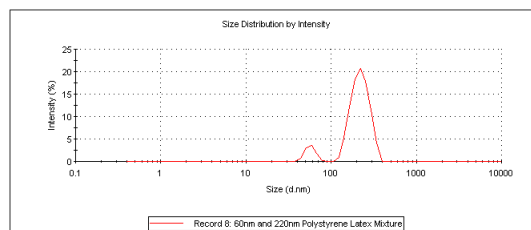


### (Modal Size Report)

- Use the **Intensity PSD** for reporting the size of each peak in the distribution
- Use the **Volume** or **Number PSD** for reporting the relative amounts of each peak in the distribution

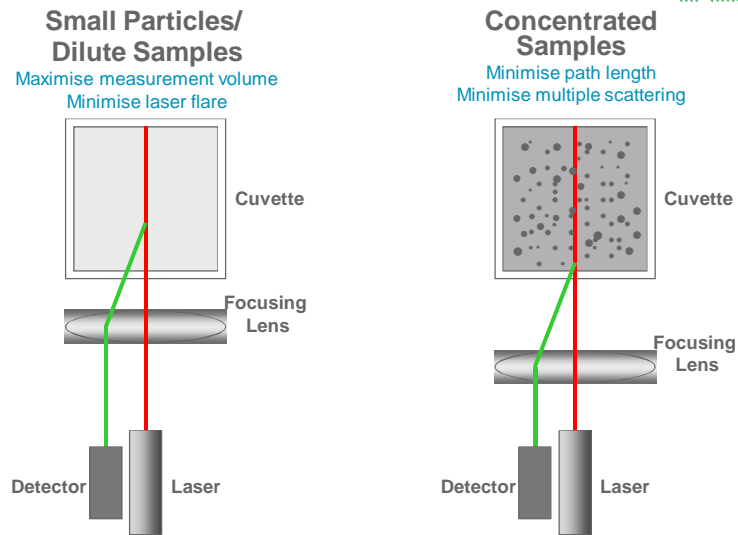
Z-Average (d<sub>z</sub>): 175.1  
 PDI: 0.191  
 Intercept: 0.866  
 Result quality: Good

Diam. (nm)	Width (nm)	% Intensity : Volume : Number		
Peak 1: 222.9	51.67	90.8	52.3	2.0
Peak 2: 57.07	7.544	9.2	47.7	98.0
Peak 3: 0.000	0.000	0.0	0.0	0.0



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## NIBS: Variable Measurement Position



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## Electrokinetic Effects



- Charged particles exhibit certain effects under the influence of an applied electric field
- Collectively defined as **electrokinetic effects**
- **Electrophoresis** = movement of a charged particle relative to the liquid it is suspended in under the influence of an applied electric field
- **Electro-osmosis** = movement of a liquid relative to a stationary charged surface under the influence of an electric field

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## Electrophoretic Light Scattering

### Zeta Potential Measurements



- Combination of **electrophoresis** and **light scattering**
- **Electrophoresis** = movement of a charged particle relative to the liquid it is suspended in under the influence of an applied electric field

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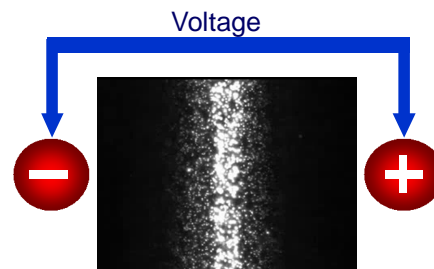
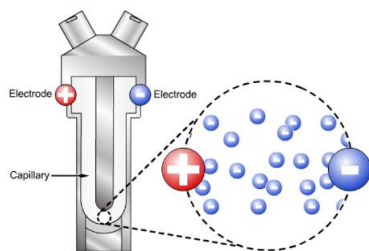


## Electrophoretic Light Scattering

### Zeta Potential Measurements



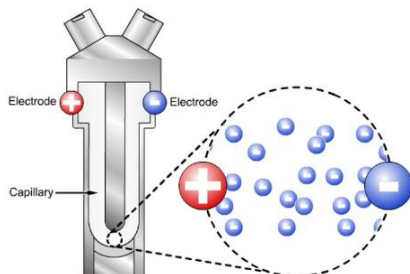
- Combination of **electrophoresis** and **light scattering**
- **Electrophoresis** = movement of a charged particle relative to the liquid it is suspended in under the influence of an applied electric field



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## Electrophoretic Light Scattering

### Zeta Potential Measurements



- Particles velocity dependent on:
  - **Zeta potential**
  - Field strength
  - Dielectric constant of medium
  - Viscosity of the medium

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## Electroosmosis



- **Electroosmosis:** the movement of a liquid relative to a stationary charged surface under the influence of an electric field

Electrophoresis in a  
Closed Capillary Cell



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## Measurement Technique M3

### Mixed Mode Measurement



- The Zetasizer Nano uses a combination of fast field reversal (**FFR**) and slow field reversal (**SFR**)
- **FFR** measures the **mean zeta potential value** due to electrophoresis only
- **SFR** allows a **zeta potential distribution** to be determined

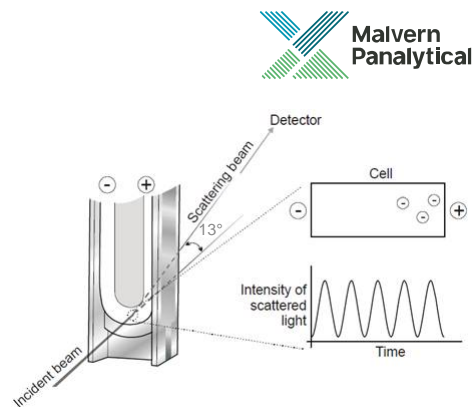
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## Electrophoretic Light Scattering

### Zeta Potential Measurements

- Scattered light is frequency shifted
- Frequency shift  

$$\Delta f = 2v \sin(\theta/2)/\lambda$$
  - $v$  = the particle velocity
  - $\lambda$  = laser wavelength
  - $\theta$  = scattering angle
- Frequency shifts determined by **Fourier transformation** and **phase analysis light scattering**
- Measured **electrophoretic mobility** converted into **zeta potential** using **Henry's equation**

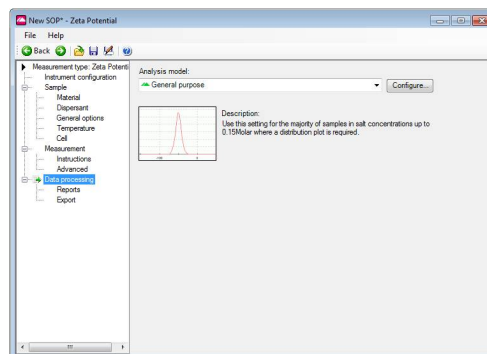


- Combination of PALS with M3 (**M3-PALS**) is patented

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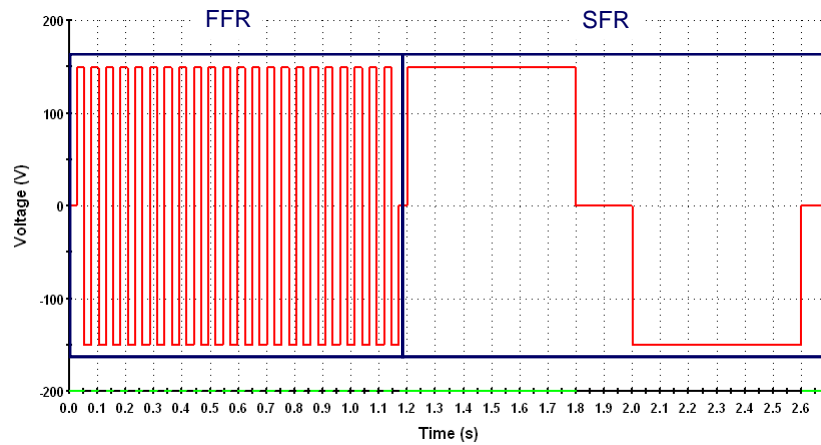
## General Purpose

- General Purpose = FFR + SFR
- FFR = zeta potential mean (electrophoresis only)
- SFR = zeta potential distribution (electrophoresis and electro-osmosis)



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## General Purpose: Voltage Plot

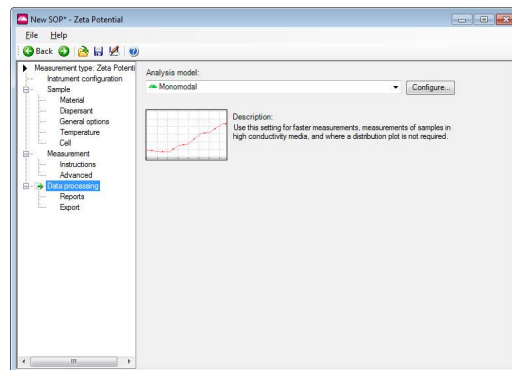


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## Monomodal

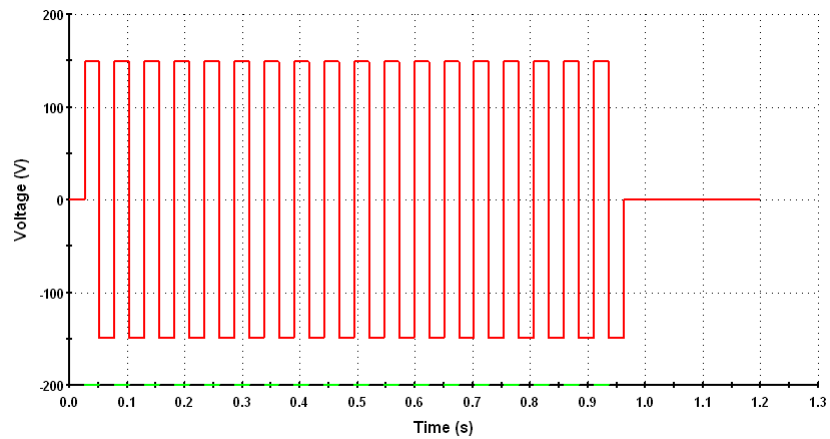


- Monomodal = FFR only
- Zeta potential mean only
- No zeta potential distribution is determined



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## Monomodal: Voltage Plot

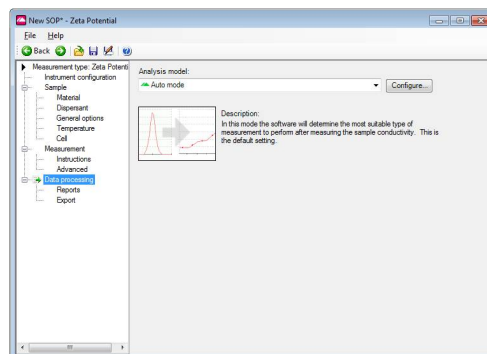


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## Auto Mode



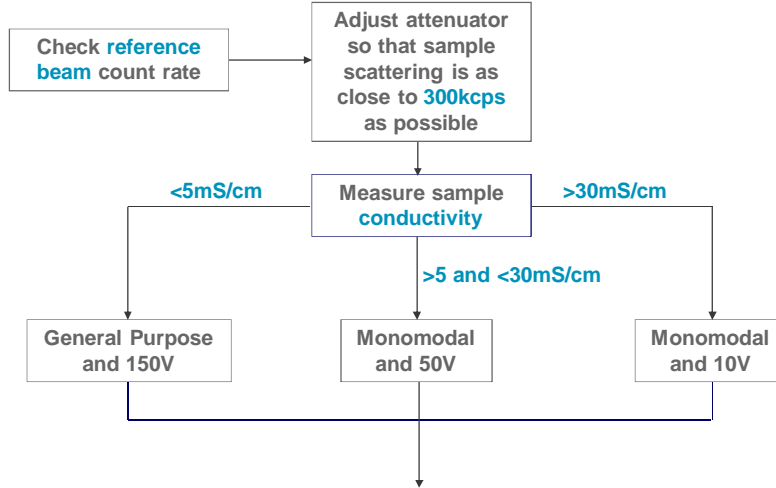
- Default measurement option
- Software determines the most suitable type of measurement to perform after measuring the sample conductivity



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### Auto Mode

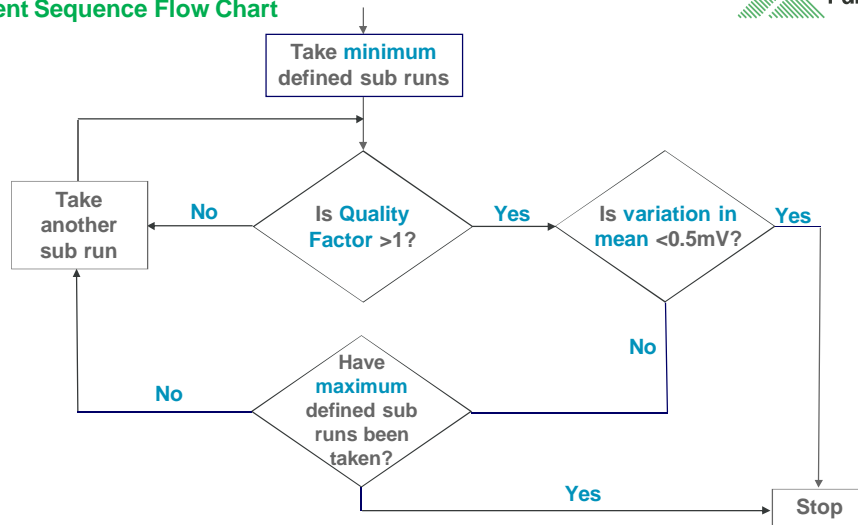
#### Measurement Sequence Flow Chart



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### Auto Mode

#### Measurement Sequence Flow Chart



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## Contents



- Instruments using light
- Light scattering
- Nanoparticle Tracking Analysis
- Dynamic light scattering (DLS)
- Electrophoretic light scattering (ELS)
- **Static light scattering (SLS)**

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## Static Light Scattering

### Absolute Molecular Weight



- $I \propto (M_w)(C)$ 
  - $I$  = intensity of scattered light
  - $M_w$  = weight-averaged molecular weight
  - $C$  = concentration
- Zetasizer measures the intensity of scattered light of various known concentrations of sample at one angle
- Called a **Debye plot** and allows for the determination of
  - Weight-averaged Molecular Weight
  - 2nd Virial Coefficient

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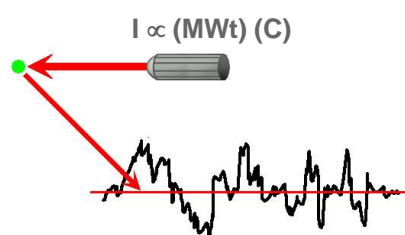
## 2nd Virial Coefficient ( $A_2$ )



- A thermodynamic property describing the interaction strength between the molecule and the solvent
- For samples where  $A_2 > 0$ , the molecules tend to stay in solution
- When  $A_2 = 0$ , the molecule-solvent interaction strength is equivalent to the molecule-molecule interaction strength – the solvent is described as being a theta solvent
- When  $A_2 < 0$ , the molecule will tend to crystallise or aggregate

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## Static Light Scattering



$$\frac{KC}{R_\theta} = \left( \frac{1}{M_w} + 2A_2C \right) P_\theta \quad (\text{Rayleigh Equation})$$

K = Optical constant  
 $M_w$  = Molecular weight  
 $A_2$  = 2<sup>nd</sup> Virial coefficient

C = Concentration  
 $R_\theta$  = Rayleigh Ratio of the sample  
 $P(\theta)$  = Shape factor

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## Static Light Scattering



$$\frac{KC}{R_\theta} = \left( \frac{1}{M_w} + 2A_2C \right) P_\theta$$

$$K = \frac{4\pi^2}{\lambda_o^4 N_A} \left( n_o \frac{dn}{dc} \right)^2$$

$\lambda_o$  = laser wavelength  
 $N_A$  = Avogadro's number  
 $n_o$  = Solvent RI  
 $dn/dc$  = differential refractive increment

$$P_\theta = 1 + \frac{16\pi^2 n_o^2 R_g^2}{3\lambda_o^2} \sin^2\left(\frac{\theta}{2}\right)$$

$R_g$  = Radius of gyration  
 $\theta$  = Measurement angle

$$R_\theta = \frac{I_A n_o^2}{I_T n_T^2} R_T$$

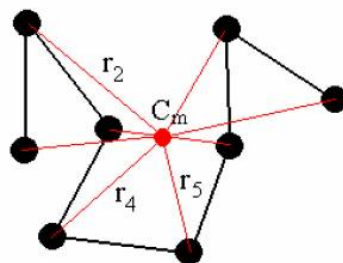
$I_A$  = Intensity of analyte (sample I – solvent I)  
 $n_o$  = Solvent RI  
 $I_T$  = Intensity of standard (toluene)  
 $n_T$  = Standard (toluene) RI  
 $R_T$  = Rayleigh ratio of standard (toluene)

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## Radius of Gyration ( $R_g$ )



Defined as the mass weighted average distance from the centre of mass to each mass element in a macromolecule



$$r_g^2 = \frac{\sum m_i r_i^2}{\sum m_i}$$

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## Static Light Scattering



$$\frac{KC}{R_{\theta}} = \left( \frac{1}{M_w} + 2A_2C \right) P_{\theta}$$

For Rayleigh scatterers,  $P(\theta) = 1$  and the equation is simplified to

$$\frac{KC}{R_{\theta}} = \left( \frac{1}{M_w} + 2A_2C \right) \quad (y = mx + c)$$

A plot of  $KC/R_{\theta}$  versus  $C$  should give a straight line whose intercept at zero concentration will be  $1/M$  and whose gradient will be  $A_2$

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## Lysozyme in PBS Example



$$\frac{dn}{dc} = 0.185 (\text{ml} / \text{g})$$

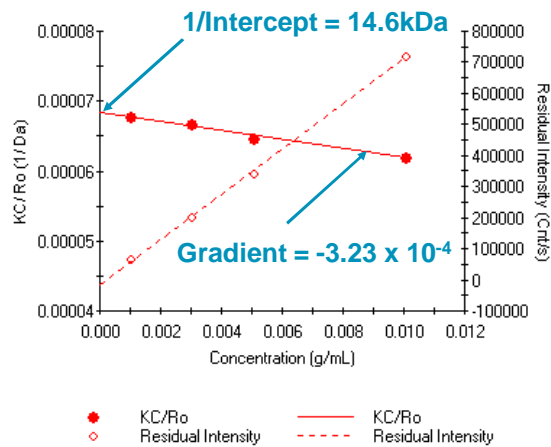
$$I_{\text{tol}} = 192630 \text{ (counts/sec)}$$

$$I_{\text{sol}} = 21870 \text{ (counts/sec)}$$

Lysozyme Conc <sup>n</sup> (mg/ml)	Measured Intensity (counts/sec)	Intensity of Analyte (counts/sec)	$KC/R_{\theta}$ (1/Da)
1.006	87,830	65,960	$6.1994 \times 10^{-5}$
3.018	222,900	201,030	$6.4765 \times 10^{-5}$
5.029	366,770	344,900	$6.6682 \times 10^{-5}$
10.059	742,570	720,700	$6.7743 \times 10^{-5}$

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## Lysozyme in PBS Example



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## Zetasizer Nano SLS $M_w$ Specifications



For single angle  $M_w$  measurements with Zetasizer Nano instruments:

- Globular proteins
  - Up to 32 nm diameter
  - Up to 20,000,000 Da (g/mol)
- Random coil polymers
  - Up to 42 nm diameter
  - Up to 500,000 Da (g/mol)

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Thank you for your attention  
Any questions?

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